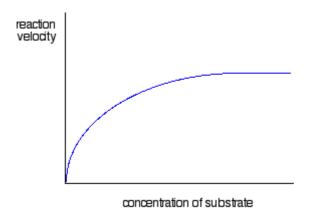
Chemguide - questions

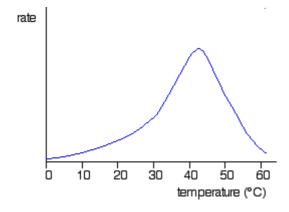
PROTEINS: CHANGING CONDITIONS IN ENZYME CATALYSIS

1. If you plot the initial rates of an enzyme-controlled reaction against the substrate concentration you get this graph (taken from the Chemguide page).



Explain the shape of the graph.

- 2. Two numbers associated with an enzyme-controlled reaction are V_{max} and K_M (the Michaelis constant). Don't waste time on this question unless you are sure this is on your syllabus.
 - a) Show on the graph how you would find the value for V_{max} .
 - b) Show on the graph how you would find the value for $K_{\rm M}$.
 - c) Suppose you had two enzymes with K_M values of 1.5 x 10^{-2} and 3.0 x 10^{-4} . Which one is working more effectively? Explain your answer.
- 3. This graph (taken from the Chemguide page) shows the effect of temperature on the rate of a typical enzyme-controlled reaction.

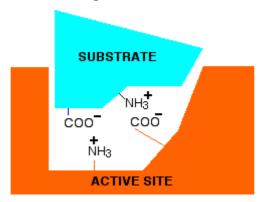


Explain the shape of the graph.

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4. Suppose an enzyme with an optimum pH of around 7 attached its substrate to its active site using the ionic attractions shown in this diagram taken from the Chemguide page.



- a) Explain what would happen if the pH of the solution was made more acidic.
- b) Explain what would happen if the pH of the solution was made more alkaline.
- c) At extreme pHs (either very acidic or very alkaline), the whole enzyme molecule might be affected not just the active site. Explain why.